

## 우엉차의 껍질 제거와 로스팅에 따른 추출온도별 항산화 효과

박미영\*† · 우등광\*\* · 박영현\*\*

\*순천향대학교 교육대학원 영양교육전공, \*\*순천향대학교 자연과학대학 식품영양학과

### Effects of Roasting and Peeling Process and Extraction Temperature on the Antioxidant Activity of Burdock Tea

Mi Young Park\*†, Chengguang Yu\*\* and Young Hyun Park\*\*

\*Department of Food and Nutrition Education, Graduate School of Education, Soonchunhyang University, Asan 31538, Korea.

\*\*Department of Food Science and Nutrition, Soonchunhyang University, Asan 31538, Korea.

#### ABSTRACT

**Background:** We investigated the optimal aqueous extraction conditions for recovery of high yields of total phenolic compounds from roots of *Arctium lappa* L. (burdock, Asteraceae), and we compared their antioxidant capacity.

**Methods and Results:** The antioxidant activity of the extracts was tested using 2,2-diphenyl-1-picrylhydrazyl, 2,2'-azino-bis (3-ethylbenzothiazoline-6-sulfonic acid)-diammonium salt, and oxygen radical absorbance capacity assays. In addition, the major phenolic compounds present in the extracts were determined by high performance liquid chromatography analysis. Our results suggest that the roasted burdock 100 °C, 15 min extract exhibited the strongest radical scavenging activity and possessed the highest concentration of phenolic compounds. The polyphenol content of both dried burdock and roasted burdock significantly increased with increase in the extraction temperature and time.

**Conclusions:** These results indicated a relationship between phenolic compound levels in burdock and their free radical scavenging activities. This suggests that phenolic compounds significantly increase the antioxidant potential of burdock extracts.

**Key Words:** *Arctium lappa* L., 2,2'-Azino-Bis (3-Ethylbenzothiazoline-6 Sulfonic Acid)-Diammonium Salt, Antioxidant Activity, 2,2-Diphenyl-1-Picrylhydrazyl, Oxygen Radical Absorbance Capacity

#### INTRODUCTION

Considerable epidemiological evidence has revealed an association between diets rich in vegetables and fruits and a decreased risk of cardiovascular diseases and certain forms of cancer (Gaté *et al.*, 1999; Yeum *et al.*, 2003). Research has shown that, these diseases occur result from a variety of exogenous and endogenous free radicals in the external environment.

The most widely used natural antioxidants are carotenoids, vitamin C, and vitamin E. These antioxidants

play a significant role in reducing excessive oxidative stresses, which leads to loss of cell function and regulation (Nordberg and Arnér, 2001). Excessive levels of reactive oxygen species (ROS) or loss of antioxidant defense systems can injure cells through lipid peroxidation, and cause DNA and protein damage (Halliwell, 1996; Shahidi *et al.*, 1992). Therefore, improving antioxidant potential is an important method to prevent the development of some chronic diseases.

Herb- and plant-based medicines contain phenolic compounds and have played an important role in human

†Corresponding author: (Phone) +82-41-530-1259 (E-mail) phy012@sch.ac.kr

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health. Accumulating evidence suggests that phenolic compounds function as reducing agents, hydrogen donors, and free-radical eliminators (Lee *et al.*, 2011; Seo *et al.*, 2009; Shahidi *et al.*, 1992). However, a lot of research is still needed to evaluate their properties and mechanisms of action.

*Arctium lappa* L. commonly known as burdock, is a perennial plant of the Asteraceae (Compositae) family. Burdock root is usually consumed as a tea and a side dish in Korea. It has been reported to have a variety of biological activities including anti-inflammatory, antiproliferative, antiviral, and antioxidant effects (da Silva *et al.*, 2013; dos Santos *et al.*, 2008; Huang *et al.*, 2010; JianFeng *et al.*, 2012; Pereira *et al.*, 2005; Wang *et al.*, 2014). Previous studies have focused on the antioxidant activity of organic solvent extracts of burdock root; however, it is unknown whether water extracts of burdock root possess antioxidant properties (da Silva *et al.*, 2013; dos Santos *et al.*, 2008; Tian *et al.*, 2014). Because burdock tea is prepared from water, it is necessary to assay the antioxidant activity of its water extract. During the preparation of burdock tea, it is dried

and roasted without being peeled in the food industry; however, the skin is often peeled following a preliminary wash in the home. Therefore, the antioxidant activity of non-peeled and peeled burdock root should be studied.

The purpose of this study was to optimize water extraction conditions for high yield total phenolic content recovery. The antioxidant capacity of water extracts of burdock root was assessed and compared using three different antioxidant assays. Additionally, the contents of major phenolic compounds in burdock root extracts were determined by high performance liquid chromatography (HPLC) analysis. These findings can be considered for the further isolation and purification of phenolic compounds from *A. lappa* root for potential use in the nutraceutical or pharmaceutical industries.

## MATERIALS AND METHODS

### 1. Sample preparation

The roots of dried- and roasted-burdock were obtained

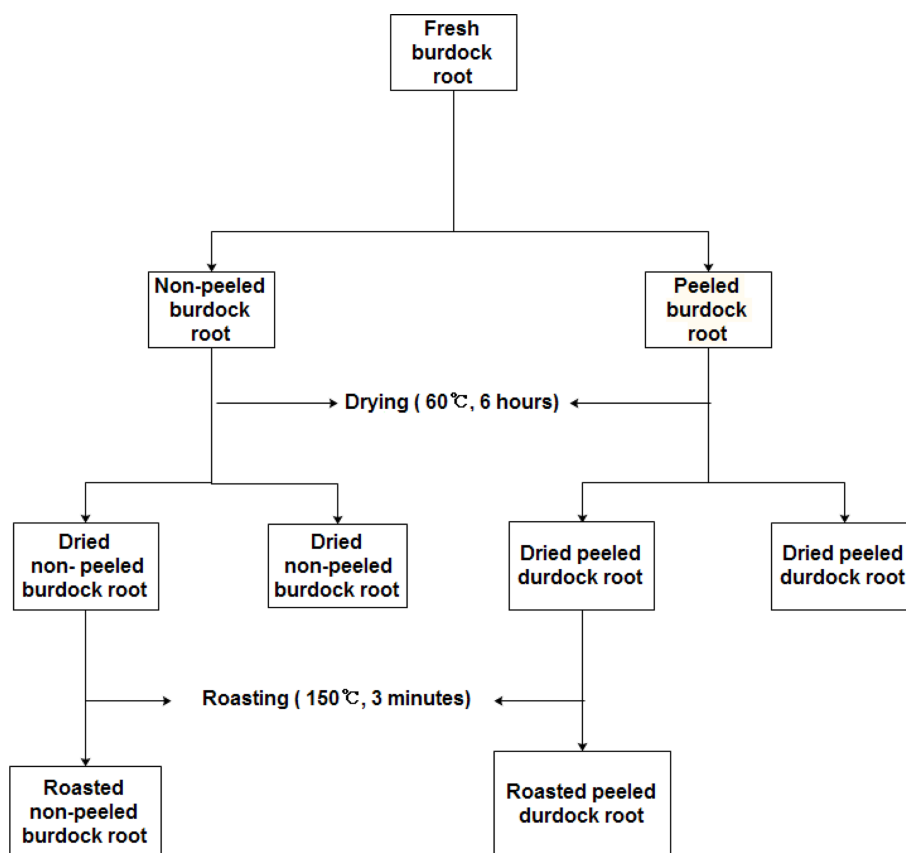


Fig. 1. The method for fresh burdock root preparation.

from Jinjusi, Gyeongsangnam-do, South Korea, and fresh roots were obtained from Asan-si, Chungcheongnam-do, South Korea. The fresh burdock roots were divided into two groups. One group contained peeled burdock roots and the other contained non-peeled burdock roots. The burdock roots were cut into thin slices (0.5 cm), and then dried for 6 h at 60°C using a drying oven (J-300M, JISICO Co., Ltd., Seoul, Korea). After drying, the two groups of burdock were divided into further two sub-groups, one of which was roasted separately in a frying-pan for 3 min at 150°C (Fig. 1). Water was selected for this study as it is safe, environmentally friendly, accessible, and cheap in comparison to the organic solvents utilized in previous studies (Vuong *et al.*, 2014).

The roots of dried burdock (DB) and roasted burdock (RB) (1 g) were extracted with water (20, 80, 100°C, 100 ml) for 5, 15, and 30 min. Furthermore, the non-peeled and peeled burdock roots (1 g) were extracted with 80°C water (100 ml) for 30 min, next, all the burdock extracts were filtered through 0.45 µm FP-Vericef™ membrane filter (VWR International, LLC., Radnor, PA, USA). All other chemicals and reagents used were purchased from Sigma-Aldrich Co. (St. Louis, MO, USA).

## 2. Total phenol determination

Total phenol contents were determined using the Folin-Ciocalteu reagent according to the method described by Ainsworth and Gillespie (2007). Briefly, 50 µl samples of various fractions were assayed with 250 µl Folin reagent and 500 µl sodium carbonate (20%, w/v). The mixture was vortexed and diluted with water to a final volume of 5 ml. After incubation for 30 min at room temperature, the absorbance was read at 734 nm using UV-VIS spectrophotometer (V-530, Jasco Co., Tokyo, Japan). Total polyphenol contents of burdock extracts were reported as mg·gallic acid equivalent/g (mg·GAE/g).

## 3. Total flavonoid determination

Total flavonoid contents were determined using the method described by Ahmed *et al.* (2014). Burdock extracts (10 mg/ml) were added to the test tube, then 2.5 ml distilled water and 75 µl 5% NaNO<sub>2</sub> solution were added, and the samples were shaken. After 5 min, 150 µl 10% AlCl<sub>3</sub> solution was added, and after 6 min, 500 µl 1 N NaOH

solution was added and the mixture was shaken. After incubation for 10 min at 37°C, the absorbance was determined at 415 nm using UV-VIS spectrophotometer (V-530, Jasco Co., Tokyo, Japan). Total flavonoid contents of burdock extracts were expressed as mg·quercetin equivalent/g (mg·QE/g).

## 4. DPPH radical scavenging activity

The antioxidant activity of the extract and fractions was determined on the basis of their ability to scavenge the stable 2,2-diphenyl-1-picrylhydrazyl (DPPH) free radical, using the method described by Braca *et al.* (2001). The absorbance at 517 nm was determined by using UV-VIS spectrophotometer (V-530, Jasco Co., Tokyo, Japan). The DPPH radical scavenging activity was calculated by the following formula:

$$\text{DPPH radical scavenging activity (\%)} = \frac{A_0 - A}{A_0} \times 100$$

where  $A_0$  is the absorbance of the control and  $A$  is the absorbance of the burdock extracts or the standard.

## 5. ABTS radical scavenging activity

The spectrophotometric analysis of 2,2'-azino-bis (3-ethylbenzothiazoline-6-sulfonic acid)-diammonium salt (ABTS) radical scavenging activity of burdock extracts was performed according to the method described by Re *et al.* (1999). The absorbance at 734 nm was determined by using UV-VIS spectrophotometer (V-530, Jasco Co., Tokyo, Japan). The ABTS<sup>•+</sup> radical scavenging activity was calculated using the following equation:

$$\text{ABTS}^{\bullet+} \text{ scavenging effect (\%)} = [1 - (A_{\text{sample}} - A_{\text{control}})] \times 100$$

where  $A_{\text{sample}}$  is the absorbance of the burdock extracts or the standard and  $A_{\text{control}}$  is the absorbance of the control.

## 6. ORAC value assay

The oxygen radical absorbance capacity (ORAC) value was determined using the modified method described by Ou *et al.* (2001). The reaction was carried out in 75 mM sodium phosphate buffer (KH<sub>2</sub>PO<sub>4</sub>-K<sub>2</sub>HPO<sub>4</sub> buffer, pH 7.4, 25 µl) and fluorescein (60 nm, 150 µl) with burdock extracts (100 µg/ml, 25 µl), 2,2'-azobis (2-methylpropanamide)

dihydrochloride (AAPH, 25  $\mu\text{l}$ ) and trolox (25  $\mu\text{l}$ ), respectively. Samples were mixed was done in black-walled 96-well plates at 37°C, and testing started immediately after mixing. Fluorescence was measured at an excitation wavelength of 485 nm and emission wavelength 535 nm every 1 min for 120 min using a microplate reader Fluostar Omega spectrophotometer (BMG Labtech, Ortenberg, Germany).

The trolox calibration curve was calculated based on the trolox concentration range of 0-50  $\mu\text{g}/\text{ml}$  (1, 5, 10, 30, and 50  $\mu\text{g}/\text{ml}$  final concentrations). The area under the curve (AUC) was calculated for each sample by integrating the relative fluorescence curve. The net AUC of the sample was calculated by subtracting the AUC of the blank. The regression equation between the AUC and the trolox concentration was determined, and ORAC values are expressed as mg · trolox equivalent/100 g (mg · TE/100 g) of samples.

### 7. Quantification of phenolic compounds by HPLC analysis

Phenolic compounds were identified and quantified by using a HPLC Spectra System SCM 1000 (Thermo Fisher Scientific Inc., Waltham, MA, USA) equipped with UV-visible detector (Shimadzu, Kyoto, Japan). Reverse phase chromatographic analyses were carried out under gradient conditions using a packed C<sub>18</sub> column (4.6 mm × 120 mm); the mobile phase was water containing 0.5% phosphoric acid and 20% ACN. The burdock extracts as well as the mobile phase were filtered through a 0.45  $\mu\text{m}$  membrane filter (Merck Millipore, Billerica, MA, USA) and then degassed in an ultrasonic bath prior to use. The extracts were analyzed for the presence or absence of gallic acid (GA), chlorogenic acid (CGA), epigallocatechin gallate (EGCG), and caffeic acid (CA). Those compounds were identified by comparing their retention time and UV absorption spectra with those of the commercial standards. The flow rate was 0.5 ml/min, injection volume was 10  $\mu\text{l}$  and the wavelength was 280 nm.

### 8. Statistical analysis

All tests were carried out in triplicate and the data are presented as means ± standard deviation (SD). Data were analyzed by One-way analysis of variance (ANOVA) followed by Duncan's Multiple Range Tests;  $p < 0.05$  was considered as statistically significant.

**Table 1.** Total polyphenol and total flavonoid contents of DB and RB extracts by different extraction method.

Sample	Contents		
	Total polyphenol (mg · GAE <sup>1</sup> /g)	Total flavonoid (mg · QE <sup>2</sup> /g)	
DB 20°C	5 min	10.86 ± 0.33 <sup>k</sup>	2.17 ± 0.09 <sup>j</sup>
	15 min	11.28 ± 0.40 <sup>k</sup>	2.60 ± 0.07 <sup>j</sup>
	30 min	12.13 ± 0.45 <sup>k</sup>	3.81 ± 0.09 <sup>i</sup>
DB 80°C	5 min	20.18 ± 0.48 <sup>l</sup>	7.67 ± 0.19 <sup>g</sup>
	15 min	25.67 ± 0.36 <sup>i</sup>	9.71 ± 0.41 <sup>ef</sup>
	30 min	26.55 ± 0.44 <sup>i</sup>	10.83 ± 0.38 <sup>e</sup>
DB 100°C	5 min	25.46 ± 0.83 <sup>l</sup>	9.28 ± 0.40 <sup>ef</sup>
	15 min	32.06 ± 0.58 <sup>h</sup>	17.12 ± 0.44 <sup>c</sup>
	30 min	38.75 ± 0.48 <sup>e</sup>	25.33 ± 0.41 <sup>b</sup>
RB 20°C	5 min	27.20 ± 0.42 <sup>hi</sup>	4.68 ± 0.41 <sup>h</sup>
	15 min	34.73 ± 0.44 <sup>g</sup>	7.60 ± 0.15 <sup>g</sup>
	30 min	36.96 ± 0.41 <sup>f</sup>	8.73 ± 0.05 <sup>f</sup>
RB 80°C	5 min	39.49 ± 0.08 <sup>e</sup>	14.31 ± 0.59 <sup>d</sup>
	15 min	45.35 ± 0.28 <sup>d</sup>	17.97 ± 0.21 <sup>c</sup>
	30 min	47.51 ± 1.16 <sup>d</sup>	19.41 ± 0.40 <sup>bc</sup>
RB 100°C	5 min	51.42 ± 1.23 <sup>c</sup>	18.67 ± 0.11 <sup>bc</sup>
	15 min	77.31 ± 0.98 <sup>b</sup>	24.41 ± 0.05 <sup>b</sup>
	30 min	83.97 ± 2.02 <sup>a</sup>	33.32 ± 0.25 <sup>a</sup>

Means values ± SD from triplicate separated experiments are shown. \*Means within a column followed by the same letter are not significant based on the DMRT (\* $p < 0.05$ ). <sup>1</sup>GAE; Gallic acid equivalents. <sup>2</sup>QE; Quercetin equivalents.

**Table 2.** Total polyphenol and total flavonoid contents of non-peeled and peeled burdock extracts.

Sample	Contents	
	Total polyphenol (mg · GAE <sup>1</sup> /g)	Total flavonoid (mg · QE <sup>2</sup> /g)
Non-peeled DB	14.33 ± 0.45 <sup>c</sup>	11.33 ± 0.17 <sup>c</sup>
Non-peeled RB	53.65 ± 1.36 <sup>a</sup>	20.09 ± 0.49 <sup>a</sup>
Peeled DB	9.47 ± 0.39 <sup>d</sup>	5.68 ± 0.19 <sup>d</sup>
Peeled RB	22.15 ± 0.66 <sup>b</sup>	13.63 ± 0.48 <sup>b</sup>

Means values ± SD from triplicate separated experiments are shown. \*Means within a column followed by the same letter are not significant based on the DMRT (\* $p < 0.05$ ). <sup>1</sup>GAE; Gallic acid equivalents. <sup>2</sup>QE; Quercetin equivalents.

## RESULTS

### 1. Total phenolic and flavonoid contents of DB and RB extracts

Plant phenolics and flavonoids are generally highly effective free radical scavengers and antioxidants (Mustafa *et al.*, 2010). The content of total phenolic compounds in the extracts from DB and RB are shown in Table 1. The total

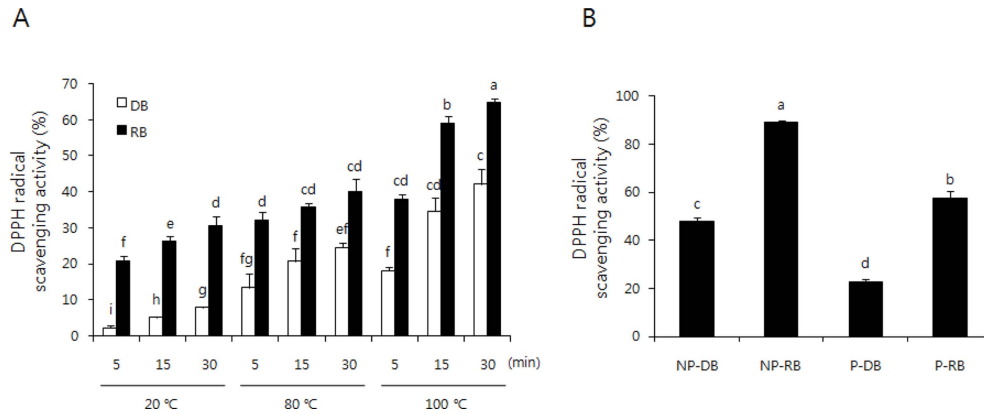
polyphenol contents of DB and RB ranged from  $10.86 \pm 0.33$  to  $38.75 \pm 0.48$ , and  $27.20 \pm 0.42$  to  $83.97 \pm 2.02$  mg·GAE/g, respectively. The total flavonoid contents of DB and RB ranged from  $2.17 \pm 0.09$  to  $25.33 \pm 0.41$  and from  $4.68 \pm 0.41$  to  $33.32 \pm 0.25$ , mg·QE/g, respectively (Table 1). Longer extraction time and higher extract temperature resulted in the extraction of higher phenolic and flavonoid compounds. In the comparison between the DB and RB extracts, using the same extraction method, the RB extracts showed higher total polyphenol contents and total flavonoid contents than the DB extracts.

In Table 2, the total phenolic and flavonoid contents of the non-peeled RB extracts were higher than those of the other extracts ( $p < 0.05$ ).

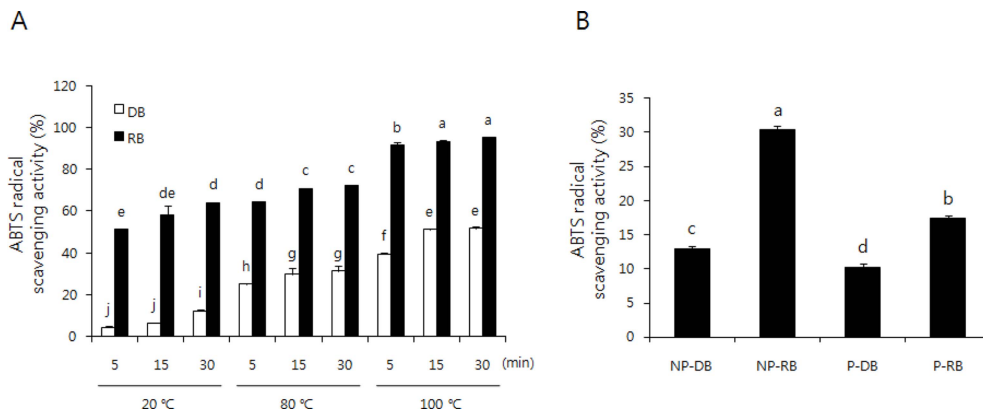
## 2. DPPH radical scavenging activity of burdock extracts

The free-radical scavenging activity of burdock extracts was assessed by DPPH assay. As shown in Fig. 2A, the RB 100 °C, 30 min extract showed the highest scavenging activity. The RB extracts showed higher DPPH radical scavenging activity than DB extracts with the same extraction method.

The DPPH radical scavenging activity of non-peeled and peeled burdock extracts decreased in the order non-peeled RB ( $88.98 \pm 0.65\%$ ) > peeled RB ( $57.53 \pm 2.74\%$ ) > non-peeled DB ( $47.82 \pm 1.50\%$ ) > peeled DB ( $22.89 \pm 0.81\%$ ). The RB extracts showed higher scavenging activity than DB extracts, and non-peeled burdock extracts showed higher scavenging activity than peeled burdock extracts (Fig. 2B).



**Fig. 2. DPPH radical scavenging activity of DB, RB, NP-B and P-B extract.** A; black or white bar represents roasted-burdock (RB) or dried-burdock (DB), respectively. B; non peeled-dried burdock (NP-DB), non peeled-roasted burdock (NP-RB), peeled-dried burdock (P-DB), peeled-roasted burdock (P-RB). Each bar represents the mean  $\pm$  SD. The bars with different letters are significantly different ( $p < 0.05$ ) from each other.



**Fig. 3. ABTS radical scavenging activity of DB, RB, NP-B and P-B extract.** A; black or white bar represents roasted-burdock (RB) or dried-burdock (DB), respectively. B; non peeled-dried burdock (NP-DB), non peeled-roasted burdock (NP-RB), peeled-dried burdock (P-DB), peeled-roasted burdock (P-RB). Each bar represents the mean  $\pm$  SD. The bars with different letters are significantly different ( $p < 0.05$ ) from each other.

### 3. ABTS radical scavenging activity of burdock extracts

The ABTS assay is based on the reaction between ABTS and potassium persulfate, which generates a blue/green ABTS radical (ABTS<sup>•+</sup>). The ABTS radical scavenging activities of various extracts increased with extract time and temperature (Fig. 3A). The ABTS radical scavenging activity of non-peeled and peeled burdock extracts decreased in the order non-peeled RB (30.43 ± 0.48%) > peeled RB (17.45 ± 0.41%) > non-peeled DB (12.99 ± 0.34%) > peeled DB (10.27 ± 0.44%). Similar to the results obtained by the DPPH radical scavenging assay, the RB extracts showed higher scavenging activity than DB extracts, and non-peeled burdock extracts showed higher ABTS radical scavenging activity than peeled burdock extracts (Fig. 3B).

### 4. ORAC value of burdock extracts

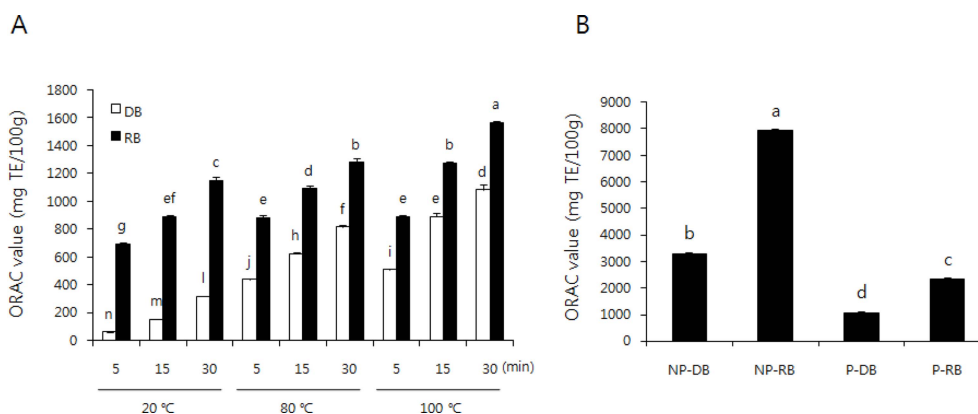
The ORAC assay is based on radical scavenging by the AAPH radical, and the ORAC has been widely employed to evaluate the antioxidant capacity of beverages, food, and plant extracts (Prior *et al.*, 2005). Similar to the DPPH and ABTS scavenging activity, obtained in the same extract time, the ORAC value increased as the temperature increased, and the RB extracts showed higher values than the DB extracts (Fig. 4A). Non-peeled burdock extracts showed higher ORAC values than peeled burdock extracts (Fig. 4B). The ORAC values of non-peeled and peeled burdock extracts decreased in the order non-peeled RB (7993.29 ± 74.90 mg · TE/100 g) > non-peeled DB (3306.03 ± 41.72 mg · TE/100 g) > peeled RB (2322.49 ±

**Table 3.** Phenolic compounds contents of DB and RB extracts by different extraction method.

Sample		Contents (mg/100 g)		
		GA <sup>1)</sup>	CGA <sup>2)</sup>	CFA <sup>3)</sup>
DB 20 °C	5 min	21.67 ± 0.09 <sup>jk</sup>	7.83 ± 0.35 <sup>n</sup>	– <sup>4)</sup>
	15 min	25.98 ± 0.32 <sup>l</sup>	11.52 ± 0.27 <sup>n</sup>	–
	30 min	31.40 ± 0.43 <sup>i</sup>	16.44 ± 0.17 <sup>m</sup>	0.32 ± 0.02 <sup>m</sup>
DB 80 °C	5 min	13.24 ± 0.44 <sup>l</sup>	107.71 ± 0.22 <sup>j</sup>	1.10 ± 0.04 <sup>l</sup>
	15 min	16.70 ± 0.17 <sup>k</sup>	129.87 ± 4.62 <sup>h</sup>	1.55 ± 0.01 <sup>l</sup>
	30 min	17.94 ± 0.50 <sup>k</sup>	136.62 ± 1.68 <sup>g</sup>	3.86 ± 0.05 <sup>k</sup>
DB 100 °C	5 min	13.63 ± 0.37 <sup>l</sup>	133.98 ± 4.52 <sup>gh</sup>	2.62 ± 0.03 <sup>kl</sup>
	15 min	21.77 ± 0.41 <sup>jk</sup>	160.78 ± 3.42 <sup>e</sup>	8.43 ± 0.29 <sup>j</sup>
	30 min	23.83 ± 0.39 <sup>j</sup>	179.81 ± 4.42 <sup>c</sup>	12.32 ± 0.32 <sup>i</sup>
RB 20 °C	5 min	74.82 ± 0.39 <sup>h</sup>	87.15 ± 3.88 <sup>l</sup>	17.35 ± 0.45 <sup>h</sup>
	15 min	96.03 ± 3.11 <sup>g</sup>	102.09 ± 2.72 <sup>k</sup>	20.57 ± 0.28 <sup>g</sup>
	30 min	116.51 ± 4.49 <sup>f</sup>	111.42 ± 2.06 <sup>i</sup>	25.04 ± 0.36 <sup>f</sup>
RB 80 °C	5 min	124.95 ± 4.76 <sup>e</sup>	127.79 ± 2.92 <sup>h</sup>	30.01 ± 0.32 <sup>e</sup>
	15 min	146.33 ± 4.29 <sup>d</sup>	145.87 ± 2.27 <sup>f</sup>	32.38 ± 0.21 <sup>d</sup>
	30 min	152.85 ± 3.28 <sup>d</sup>	174.43 ± 3.14 <sup>d</sup>	36.40 ± 0.39 <sup>c</sup>
RB 100 °C	5 min	170.82 ± 3.71 <sup>c</sup>	158.56 ± 3.16 <sup>e</sup>	33.14 ± 0.69 <sup>d</sup>
	15 min	218.73 ± 3.66 <sup>b</sup>	222.25 ± 3.86 <sup>b</sup>	38.46 ± 1.01 <sup>b</sup>
	30 min	257.13 ± 4.25 <sup>a</sup>	242.16 ± 4.29 <sup>a</sup>	45.79 ± 2.03 <sup>a</sup>

Means values ± SD from triplicate separated experiments are shown. \*Means within a column followed by the same letter are not significant based on the DMRT (\**p* < 0.05). <sup>1)</sup>GA; Gallic acid, <sup>2)</sup>CGA; Chlorogenic acid, <sup>3)</sup>CFA; Caffeic acid, <sup>4)</sup>–; no detectable.

76.94 mg · TE/100 g) > peeled DB (1,081.85 ± 40.90 mg · TE/100 g). These data indicate that the presence of phenolic and flavonoid compounds in extracts contributes significantly to their antioxidative potential.



**Fig. 4.** ORAC value of DB, RB, NP-B and P-B extract. A; black or white bar represents roasted-burdock (RB) or dried-burdock (DB), respectively. B; non peeled-dried burdock (NP-DB), non peeled-roasted burdock (NP-RB), peeled-dried burdock (P-DB), peeled-roasted burdock (P-RB). Each bar represents the mean ± SD. The bars with different letters are significantly different (*p* < 0.05) from each other.

## 5. Quantification of phenolic compounds by HPLC analysis

Previous studies have identified GA, CGA, and CA in burdock roots (Tezuka *et al.*, 2013; Saleem *et al.*, 2009). As the extract temperature and time were increased, the polyphenol contents of both DB and RB significantly increased (Table 3).

## DISCUSSION

Burdock (*Arctium lappa* L.) root has been widely used as a folk medicine and vegetable in East Asia. In the present study, we investigated the optimal water extraction conditions for high-yield total phenolic content recovery, and the antioxidant capacity of the water extracts of *A. lappa* root.

In the DPPH radical scavenging assay, ABTS radical scavenging assay, and ORAC assay, the RB 100°C, 30 min extract showed the highest antioxidant activity, and non-peeled burdock extracts showed higher scavenging activity than peeled burdock extracts (Fig. 2, 3, 4). The ABTS radical scavenging activity of RB (80-100°C) is almost two-fold higher than that of DB (Fig. 3). As expected, the total phenolic acid and flavonoid contents of the RB 100°C, 30 min sample were higher than those of the other extracts (Table 1). These results indicate that there is a relationship between the phenolic compound concentrations in burdock and their free radical scavenging activities. Therefore, the presence of phenolic compounds significantly contributes to their antioxidant potential. These data are consistent with those from previous studies showing that a highly positive relationship exists between total phenolic contents and antioxidant activity in many plants (Chukwumah *et al.*, 2009; GURSOY *et al.*, 2009).

The phenolic and flavonoids compound contents of non-peeled burdock extracts are greater than those of peeled burdock extracts (Table 2). These results support the conclusions of others studies showing that fruits and vegetable peels possess a higher content of these compounds than pulps (Asikin *et al.*, 2012; Gorinstein *et al.*, 2001).

Although caffeoylquinic acids, dicaffeoylquinic acids, CA derivatives, hydroxycinnamoylquinic acids, and CGA have been identified in burdock roots (Jaiswal and Kuhnert, 2011; Lin and Harnly, 2008; Liu *et al.*, 2012; Saleem *et al.*, 2009), most studies analyzed the organic solvent

extract of burdock roots. Since burdock root has been consumed as a tea, we analyzed phenolic compounds in the water extract of burdock roots. As the extract temperature and time increased, the contents of phenolic compounds also increased (Table 3).

The CGA content was highest among the three phenolic compounds in DB (Table 3). The GA content increased as much as the CGA content in RB. The RB 100°C, 30 min extract contained 10-fold more CGA than the DB 100°C, 30 min extract (Table 3). Roasting treatment led to a significant increase in the level of GA, CGA, and CA, which correlates well with the observed antioxidant activity of burdock root extract. GA is a strong antioxidant compound and is used as a reference in the assessment of antioxidant activity. Its oxidation involves 4.6 electrons, whereas ascorbic acid involves two electrons (Hotta *et al.*, 2002). CGA is the ester of CA and quinic acid and also possesses antioxidant properties (Sato *et al.*, 2011; Xu *et al.*, 2012). During the roasting process, GA may be released from tannin and CA may be generated from CGA degradation.

Thus, the antioxidant activity of the RB 100°C, 30 min extract is partly due to the presence of those phenolic compounds. The results indicate that the roasting process and conditions play an important role in the phenolic compound content of burdock tea. In addition to burdock root phenolic compounds, water-soluble polysaccharide, fructo-oligosaccharide, arctigenin and inulin are also known to have antioxidant activity (Li *et al.*, 2008; Liu *et al.*, 2014; Zhao *et al.*, 2009). The antioxidant activity of burdock root may involve a synergistic effect among phytochemicals.

In this study, an assay of antioxidant activity was tested for the first time and the total phenolic and flavonoid contents of water extracts from burdock root were determined. Antioxidant and radical scavenging ability of the extracts were found to be extract temperature and time dependent.

A key finding of this study was that the RB 100°C, 30 min extract exhibited the strongest radical scavenging activity and possessed the highest concentration of phenolic compounds. Unfortunately, we did not analyze the contents of various phenolic compounds or those in non-peeled and peeled burdock extracts. However, the non-peeled RB extract was found to have a higher ORAC value than

peeled RB extract, suggesting that non-peeled burdock extract has more antioxidant benefits than peeled burdock extract. Roasting treatment generated a high amount of phenolic compounds, in particular, GA.

Thus, non-peeled RB can be very beneficial for enhanced phenolic compound extraction with desirable benefits. Future studies should focus on the roasting condition, the identification of these antioxidants, and the purification of this plant ingredient to obtain agents with high efficacy and activity.

## ACKNOWLEDGEMENT

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